

2,4-Diamino-5-(3,4,5-trimethoxybenzyl)-
pyrimidinium 4-hydroxy-3-nitrophenyl-
arsonate monohydrateTian-Tian Pan,^a Bing-Xin Liu^b and
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The crystal structure of the title compound, (C₁₄H₁₉N₄O₃)[AsO₂(OH)(C₆H₄NO₃)]·H₂O, consists of substituted pyrimidine cations, roxarsone (3-nitro-4-hydroxyphenylarsonate) anions and water molecules. Due to deprotonation of the arsonate group, the As—O bond distances in the anion are significantly different. An extensive network of O—H···O, N—H···O and N—H···N hydrogen bonds helps to stabilize the crystal structure.

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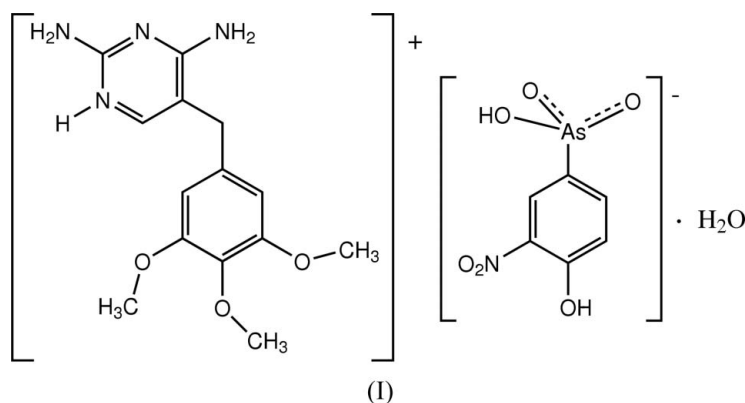
Key indicators

Single-crystal X-ray study
T = 294 K
Mean σ (C—C) = 0.006 Å
Disorder in solvent or counterion
R factor = 0.057
wR factor = 0.119
Data-to-parameter ratio = 12.9

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

Comment

3-Nitro-4-hydroxyphenylarsonic acid (common name roxarsone) is an arsenic-containing compound that has anticoccidial action and promotes growth in animals. As part of our studies into its bioactivity, we have prepared the title roxarsionate-containing molecular salt, (I), and its crystal structure is presented here.



The crystal structure of (I) consists of substituted pyrimidinium cations, roxarsionate anions and water molecules of crystallization (Fig. 1). Within the roxarsone monoanion, the As—O₂ bond distance is almost identical to that of As—O₃, but significantly shorter than the As—O₁ bond distance (Table 1), due to deprotonation of the arsonate group. The O atoms of the nitro group are disordered over two sites. Both conformations are tilted with respect to the C1-containing benzene ring plane, with dihedral angles of 27 (1) and 35 (2)° for the O5a- and O5b-nitro groups, respectively.

Within the substituted pyrimidine cation, the C12—C17 benzene ring is roughly perpendicular to the pyrimidine ring, the dihedral angle being 79.74 (11)°. The C atoms of the C18- and C19-methyl groups are coplanar with the benzene plane, their atomic deviations being 0.040 (8) and 0.054 (10) Å, respectively. Conversely, the C20-methyl C atom is out of the benzene plane by 1.248 (7) Å, thus minimizing the steric repulsion between adjacent —CH₃ groups.

An extensive network of O—H···O, N—H···O and N—H···N hydrogen bonds (Table 2) helps to stabilize the crystal structure of (I).

Experimental

2,4-Diamino-5-(3,4,5-trimethoxybenzyl)pyrimidine (5 mmol) and 3-nitro-4-hydroxyphenylarsonic acid (5 mmol) were dissolved in a water–ethanol mixture (50 ml, 4:1). The solution was stirred for 2 h at 323 K and then filtered. Single crystals of (I) were obtained from the filtrate after 3 weeks.

Crystal data

(C₁₄H₁₉N₄O₃)[AsO₂(OH)·(C₆H₄NO₃)·H₂O]
M_r = 571.38
 Orthorhombic, *Pbca*
a = 8.1865 (13) Å
b = 20.9285 (16) Å
c = 28.304 (2) Å
V = 4849.3 (9) Å³

Z = 8
D_x = 1.565 Mg m⁻³
 Mo *K*α radiation
 μ = 1.47 mm⁻¹
T = 294 (2) K
 Prism, yellow
 0.34 × 0.30 × 0.25 mm

Data collection

Bruker APEXII diffractometer 24891 measured reflections
 φ and ω scans 4489 independent reflections
 Absorption correction: multi-scan 3922 reflections with *I* > 2σ(*I*)
SADABS (Sheldrick, 2002) *R*_{int} = 0.046
*T*_{min} = 0.622, *T*_{max} = 0.700 θ _{max} = 25.5°

Refinement

Refinement on *F*²
R[*F*² > 2σ(*F*²)] = 0.057
wR(*F*²) = 0.119
S = 1.21
 4489 reflections
 347 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0447P)^2 + 5.6729P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$
 $\Delta\rho_{\max} = 0.64 \text{ e \AA}^{-3}$
 $\Delta\rho_{\min} = -0.51 \text{ e \AA}^{-3}$

Table 1

Selected bond lengths (Å).

As—O1	1.720 (3)	As—O3	1.652 (3)
As—O2	1.664 (3)	As—C1	1.903 (3)

Table 2

Hydrogen-bond geometry (Å, °).

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
N2—H2 <i>A</i> ···O3	0.95	1.87	2.817 (4)	176
N2—H2 <i>B</i> ···N4 ⁱ	0.96	2.48	3.318 (5)	146
N3—H3 <i>A</i> ···O7 ⁱⁱ	0.92	2.51	3.223 (4)	135
N3—H3 <i>B</i> ···O4 ⁱⁱⁱ	0.92	2.33	3.139 (5)	147
N3—H3 <i>B</i> ···O5 <i>A</i> ⁱⁱⁱ	0.92	2.48	3.078 (10)	123
N5—H5 <i>A</i> ···O3 ⁱ	0.94	1.70	2.607 (5)	162
O1—H1···O2 ^{iv}	0.93	1.67	2.582 (4)	165
O4—H4 <i>A</i> ···O1 <i>W</i>	0.92	1.65	2.545 (4)	165
O1 <i>W</i> —H1 <i>A</i> ···O2 ^v	0.86	1.83	2.687 (4)	171
O1 <i>W</i> —H1 <i>B</i> ···O8 ^{vi}	0.89	2.46	3.221 (4)	143
O1 <i>W</i> —H1 <i>B</i> ···O9 ^{vi}	0.89	2.01	2.767 (4)	142

Symmetry codes: (i) $x + \frac{1}{2}, -y + \frac{1}{2}, -z$; (ii) $x - \frac{1}{2}, y, -z + \frac{1}{2}$; (iii) $x - 1, y, z$; (iv) $-x, -y, -z$; (v) $x + 1, y, z$; (vi) $-x + 1, y - \frac{1}{2}, -z + \frac{1}{2}$.

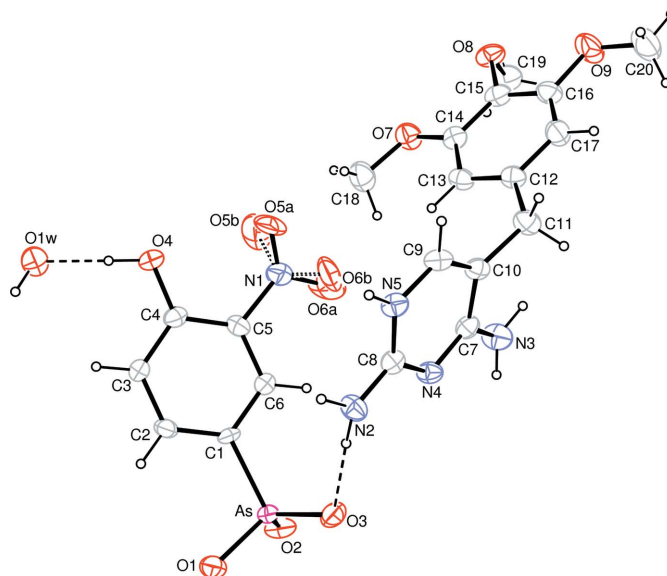


Figure 1

The molecular structure of (I) with 30% probability displacement ellipsoids (arbitrary spheres for H atoms). Dashed lines indicate hydrogen bonds. Suffixes a and b indicate the two disordered components of the nitro group.

The nitro group is disordered over two sites; the occupancies of the O atoms were refined and converged to 0.60 (3) for O5*a*/O6*a* and 0.40 (3) for O5*b*/O6*b* (occupancy sum constrained to unity). H atoms bonded to N and O atoms were located in a difference Fourier map and refined as riding in their as-found relative positions, with $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{O}, \text{N})$ [N—H = 0.92–0.96 Å and O—H = 0.92–0.93 Å]. Methyl H atoms were placed in calculated positions, with C—H = 0.96 Å, and their torsion angles were refined to fit the electron density, with $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$. Other H atoms were placed in calculated positions, C—H = 0.93 Å (aromatic) or 0.97 Å (methylene), and refined in riding mode, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

Data collection: *SMART* (Bruker, 2004); cell refinement: *SAINTE* (Bruker, 2004); data reduction: *SAINTE*; program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1993); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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References

- Altomare, A., Casciarano, G., Giacovazzo, C. & Guagliardi, A. (1993). *J. Appl. Cryst.* **26**, 343–350.
 Bruker (2004). *SMART* (Version 6.36A) and *SAINTE* (Version 7.12A). Bruker AXS Inc., Madison, Wisconsin, USA.
 Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
 Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.
 Sheldrick, G. M. (1997). *SHELXL97*. University of Göttingen, Germany.
 Sheldrick, G. M. (2002). *SADABS*. Version 2.03. University of Göttingen, Germany.